

СИНТЕЗ ТА АНАЛІЗ БІОЛОГІЧНО АКТИВНИХ РЕЧОВИН

Recommended by Doctor of Pharmacy, professor S.V.Kolisnyk

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SYNTHESIS AND THE ANTIMICROBIAL ACTIVITY OF 5-METHYL-6-(2-METHYL-1,3-THIAZOL-4-YL)-3- PHENYLTHIENO[2,3-d]PYRIMIDINE-2,4(1H,3H)-DIONES

S.V.Vlasov, T.P.Osolodchenko, S.M.Kovalenko, V.P.Chernykh

National University of Pharmacy

State Institution “Institute of Microbiology and Immunology named after I.I.Mechnikov of the National Academy of Medical Sciences of Ukraine”

Key words: thiophene; pyrimidine; alkylation; thiazole

*By the reaction of 3-phenyl-6-(α -bromoacetyl)-5-methylthieno[2,3-d]pyrimidine-2,4(1H,3H)-dione with thioacetamide 5-methyl-6-(2-methyl-1,3-thiazol-4-yl)-3-phenylthieno[2,3-d]pyrimidine-2,4(1H,3H)-dione was obtained; further the compound was modified by alkylation of its position 1 with benzyl chlorides and chloroacetamides. The structures of the compounds obtained have been confirmed by ^1H NMR and mass-spectral data. All the ^1H NMR spectra of the compounds obtained have the signals of thiazole cycle at 7.62-7.57 ppm, for 1-alkyl-5-methyl-6-(2-methyl-1,3-thiazol-4-yl)-3-phenylthieno[2,3-d]pyrimidine-2,4(1H,3H)-diones the signals of the methylene group protons are observed in the range of 5.13-5.21 ppm for benzyl substituted derivatives and 4.78-4.82 ppm for the compounds with the acetamide fragment; the last ones also contain the sharp signal of NH proton in the range of 9.68-10.41 ppm. 5-Methyl-6-(2-methyl-1,3-thiazol-4-yl)-3-phenylthieno[2,3-d]pyrimidine-2,4(1H,3H)-dione showed the antimicrobial activity against *Staphylococcus aureus* higher than the reference drugs Metronidazole and Streptomycin, it also appeared to be moderately active against *Pseudomonas aeruginosa* and *Candida albicans* fungi. The antimicrobial activity of 1-alkyl-5-methyl-6-(2-methyl-1,3-thiazol-4-yl)-3-phenylthieno[2,3-d]pyrimidine-2,4(1H,3H)-diones is inferior to the activity of the compound with the hydrogen atom in position 1; the highest activity has been determined for the derivative with 4-methylbenzyl substituent in position 1, which inhibits the growth of *Staphylococcus aureus* and *Candida albicans*.*

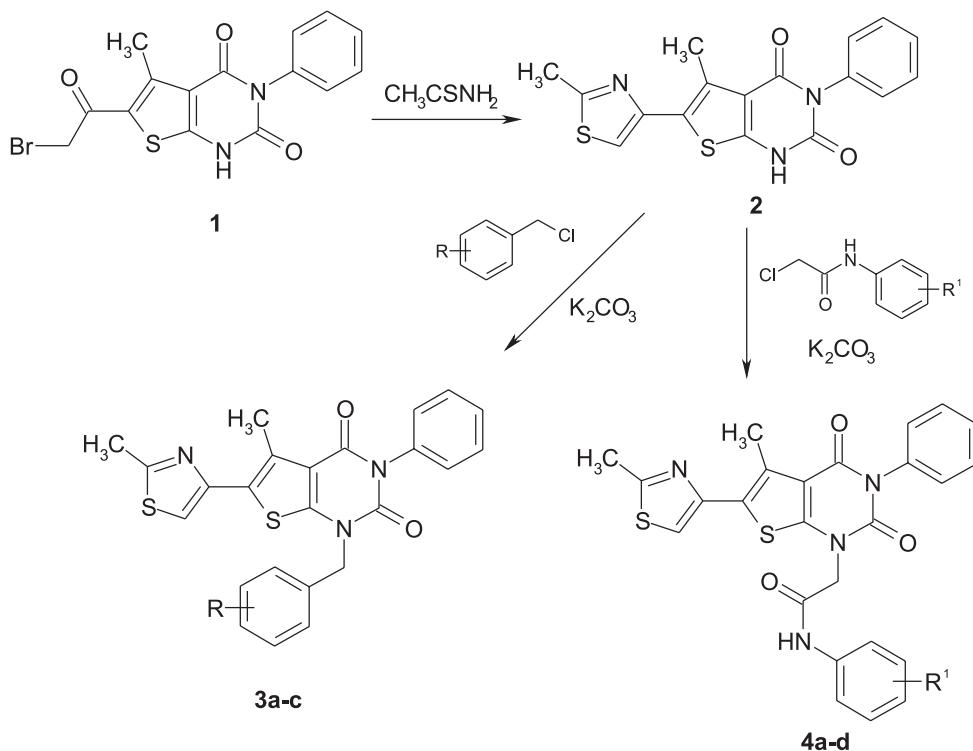
In the last years the interest to derivatives of 6-hetarylthieno[2,3-d]pyrimidines has grown. They were reported as adenosine A_{2A} receptors antagonists [8], as the compounds with antioxidant [10] activity and acetyl-CoA carboxylase (ACC) inhibitors [7]. Some compounds of the similar structure are potential anti-viral agents [9]. Our last investigations also confirm the expediency of searching novel antimicrobial agents in the range of thieno[2,3-d]pyrimidine derivatives modified in position 6 with the aromatic heterocyclic substituents [2, 3, 4, 12]. We have also found that in some cases the presence of the electron-withdrawing substituent in position 6 of thieno[2,3-d]pyrimidine together with alkylation of the nitrogen atom in position 1 promotes the antimicrobial activity of the compounds [11].

In order to construct the novel 6-hetarylthieno[2,3-d]pyrimidine-2,4(1H,3H)-diones we performed the reaction of previously reported 3-phenyl-6-(α -bromoacetyl)-5-methylthieno[2,3-d]pyrimidine-2,4(1H,3H)-dione **1** [4] with thioacetamide in the acetic acid medium and as the result 5-methyl-6-(2-methyl-1,3-thiazol-4-yl)-3-phenylthieno[2,3-d]pyrimidine-2,4(1H,3H)-dione **2** was obtained.

Further alkylation of compound **2** was performed with benzyl chlorides and chloroacetamides in the DMF medium promoted with potassium carbonate (Scheme). The series of 1-alkyl-5-methyl-6-(2-methyl-1,3-thiazol-4-yl)-3-phenylthieno[2,3-d]pyrimidine-2,4(1H,3H)-diones **3** and **4** were isolated as white crystalline solids (Tab. 1).

In the ^1H NMR spectrum of compound **2** the signals of thiazole and thiophene methyl groups are very close to each other 2.57 (3H, s, CH₃) and 2.67 (3H, s, CH₃); the signal of aromatic proton of thiazole is observed at 7.57 ppm. In all of the ^1H NMR spectra of compounds **3** and **4** the signals of the methylene group protons are observed in the range of 5.13-5.21 ppm for compounds **3**, and 4.78-4.82 ppm for derivatives **4**. The sharp signals of acetamide NH are present in the spectra of compounds **4** in the region of 9.68-10.41 ppm.

For all of the compounds obtained the screening of the antimicrobial activity by the agar diffusion method has been performed; the results are given in Tab. 3. The screening showed that 5-methyl-6-(2-methyl-1,3-thiazol-4-yl)-3-phenylthieno[2,3-d]pyrimidine-2,4(1H,3H)-dione **2** displayed the highest antimicrobial activity; its acti-



Scheme

vity against *Staphylococcus aureus* was higher than that of the reference drugs Metronidazole and Streptomycin; it also appeared to be moderately active against *Pseudomonas aeruginosa* and *Candida albicans* fungi. Alkylation of position 1 of the thieno[2,3-*d*]pyrimidine-2,4(1*H*,3*H*)-dione fragment decreases the antimicrobial activity.

Among 1-alkyl-5-methyl-6-(2-methyl-1,3-thiazol-4-yl)-3-phenylthieno[2,3-*d*]pyrimidine-2,4(1*H*,3*H*)-diones **3** and **4** the highest activity was revealed by derivative **3b** with the 4-methylbenzyl substituent in position 1, which moderately inhibited the growth of *Staphylo-*

coccus aureus and *Candida albicans*. It is noteworthy that none of the compounds tested inhibited the growth of *Bacillus subtilis* strain.

Materials and Methods Chemical Part

The melting points (°C) were measured with a Kofler melting point apparatus and were not corrected. The IR spectrum was recorded on a Bruker Tensor 27 spectrometer in KBr. ^1H NMR spectra were recorded on a Varian Mercury (200 MHz) spectrometer in $\text{DMSO}-d_6$ using TMS as an internal standard (chemical shifts are in ppm). LC/MS was recorded with a PE SCIEX API

Table 1

Physicochemical properties of 1-alkyl-5-methyl-6-(2-methyl-1,3-thiazol-4-yl)-3-phenylthieno[2,3-*d*]pyrimidine-2,4(1*H*,3*H*)-diones **3** and **4**

Compd. No.	R	R^1	Mol. formula M.w.	Yield %, in the alkylation step	M.p., °C	N%
						calc. found
3a	H	–	$\text{C}_{24}\text{H}_{19}\text{N}_3\text{O}_2\text{S}_2$ 445.57	76	279-280	9.43 9.67
3b	4-Me	–	$\text{C}_{25}\text{H}_{21}\text{N}_3\text{O}_2\text{S}_2$ 459.59	83	272-274	9.14 9.18
3c	4-F	–	$\text{C}_{24}\text{H}_{18}\text{FN}_3\text{O}_2\text{S}_2$ 463.56	71	253-255	9.06 9.32
4a	–	4-Me	$\text{C}_{26}\text{H}_{22}\text{N}_4\text{O}_3\text{S}_2$ 502.62	71	> 300	11.15 11.29
4b	–	2,4-diMe	$\text{C}_{27}\text{H}_{24}\text{N}_4\text{O}_3\text{S}_2$ 516.65	79	>300	10.84 10.97
4c	–	4-OEt	$\text{C}_{27}\text{H}_{24}\text{N}_4\text{O}_4\text{S}_2$ 532.64	86	>300	10.52 10.79
4d	–	3,5-diOMe	$\text{C}_{27}\text{H}_{24}\text{N}_4\text{O}_5\text{S}_2$ 548.64	77	293-295	10.21 10.32

Table 2

Data of ^1H NMR-spectra of 1-alkyl-5-methyl-6-(2-methyl-1,3-thiazol-4-yl)-3-phenylthieno[2,3-*d*]pyrimidine-2,4(*1H,3H*)-diones 3 and 4

Compnd. No.	Chemical shift, δ , ppm.			
	NH	Thiophene CH_3 (3H, s.) Thiazole CH_3 (3H, s.)	Aliphatic protons	Aromatic protons
3a	–	2.52 + 2.64	5.21 (2H, s, CH_2);	7.25-7.53 (10H, m, Ar-H); 7.62 (1H, s, CH thiazole)
3b	–	2.58 + 2.64	2.26 (3H, s, CH_3); 5.13 (2H, s, CH_2);	7.1-7.5 (9H, m, Ar-H); 7.62 (1H, s, CH thiazole)
3c	–	2.52 + 2.64	5.13 (2H, s, CH_2);	7.05-7.5 (9H, m, Ar-H); 7.62 (1H, s, CH thiazole)
4a	10.21 (1H, s)	2.64+2.66	2.23 (3H, s, CH_3); 4.80 (2H, s, CH_2);	7.0-7.5 (9H, m, Ar-H); 7.67 (1H, s, CH thiazole)
4b	9.68 (1H, s)	2.63+2.66	2.15 (3H, s, CH_3); 2.22 (3H, s, CH_3); 4.82 (2H, s, CH_2);	6.8-7.5 (9H, m, Ar-H); 7.67 (1H, s, CH thiazole)
4c	10.18 (1H, s)	2.64+2.66	1.28 (3H, t, CH_3); 3.97 (2H, q, CH_2); 4.78 (2H, s, CH_2);	6.86 (2H, d, Ar-H); 7.27 (2H, d, Ar-H); 7.46 (5H, m, Ar-H); 7.64 (1H, s, CH thiazole)
4d	10.41 (1H, s)	2.63+2.66	3.67 (6H, s, 2OCH_3); 4.80 (2H, s, CH_2);	6.23 (1H, m, Ar-H); 6.79 (2H, d, Ar-H); 7.27 (2H, d, Ar-H); 7.44 (3H, m, Ar-H); 7.65 (1H, s, CH thiazole)

150EX chromatograph equipped with a mass-spectrometer using the column C18 (100×4 MM), the cycle of analysis was 25 min.

The method for preparation of 5-methyl-6-(2-methyl-1,3-thiazol-4-yl)-3-phenylthieno[2,3-*d*]pyrimidine-2,4(*1H,3H*)-dione 2

To 2 g of thioacetamide in glacial acetic acid add 9 g of 3-phenyl-6-(α -bromoacetyl)-5-methylthieno[2,3-*d*]py-

rimidin-2,4(*1H,3H*)-dione 1 and reflux the mixture till formation of a blue-violet precipitate and then additionally for 2-3 hours. After cooling dilute the mixture with water, filter the precipitate formed. Next suspend the precipitate in water, alkalinify with the concentrated ammonia solution, and boil the suspension for 1 hour. Filter the precipitate formed, wash with plenty of water and dry.

Table 3

The antimicrobial activity of 5-methyl-6-(2-methyl-1,3-thiazol-4-yl)-3-phenylthieno[2,3-*d*]pyrimidine-2,4(*1H,3H*)-diones 2, 3 and 4

Compnd. No.	Diameter of the growth inhibition zone*, mm					
	Staphylococcus aureus ATCC 25923	Escherichia coli ATCC 25922	Proteus vulgaris ATCC 4636	Pseudomonas aeruginosa ATCC 27853	Bacillus subtilis ATCC 6633	Candida albicans ATCC 653/885
2	+++	+	+	+	–	++
3a	–	–	–	–	–	–
3b	++	+	+	+	–	++
3c	–	–	–	–	–	–
4a	+	–	–	–	–	–
4b	–	–	–	–	–	+
4c	–	–	–	–	–	–
4d	+	–	–	–	–	+
Metr. **	+	+	–	–	++	+
Strept. ***	++	++	–	–	++	–

* “–” – diameter of the growth inhibition zone is less than 10 mm; “+” – diameter of the growth inhibition zone is 10-15 mm;

“++” – diameter of the growth inhibition zone is 15-20 mm; “+++” – diameter of the growth inhibition zone is more than 20 mm;

** Metr. – Metronidazole DMSO solution (the concentration is 30 $\mu\text{g}/\text{ml}$);

*** Strept. – Streptomycin H_2O solution (the concentration is 30 $\mu\text{g}/\text{ml}$).

M.p. > 300°C.

Yield – 75%.

IR (KBr): 3435, 3191, 3062, 2911, 1719, 1654, 1567, 1529, 1503, 1454, 1437, 1375, 1352, 1297, 1273, 1173, 1155, 1121, 1058, 1028, 986, 961, 911, 851, 766, 734, 711, 700, 686, 664, 617, 579, 535, 495 cm⁻¹.

¹H NMR (200 MHz, DMSO-*d*₆): 2.57 (3H, s, CH₃), 2.67 (3H, s, CH₃), 7.27 (2H, m, Ar-H), 7.42 (3H, m, Ar-H), 7.57 (1H, s, CH thiazole), 12.36 (1H, br. s, NH).

LC/MS: m/z (MH⁺) 356.

Found, %: N 11.99. C₁₇H₁₃N₃O₂S₂. Calculated, %: N 11.82. M. 355.44.

The general method for synthesis of 1-alkyl-5-methyl-6-(2-methyl-1,3-thiazol-4-yl)-3-phenylthieno[2,3-*d*]pyrimidine-2,4(1*H*,3*H*)-diones 3 and 4.

To the suspension of 0.15 g of 5-methyl-6-(2-methyl-1,3-thiazol-4-yl)-3-phenylthieno[2,3-*d*]pyrimidine-2,4(1*H*,3*H*)-dione and 0.06 g of potassium carbonate add 0.00045 mole of the corresponding alkylating agents and stir the mixture when heating (60–80°C) for 5–7 hours. After cooling dilute the reaction mixture with water and filter the precipitate formed.

The study of the antimicrobial activity

According to the WHO recommendations [1, 5, 6] the following test-strains have been used: *Staphylococcus aureus* ATCC 25923, *Escherichia coli* ATCC 25922, *Pseudomonas aeruginosa* ATCC 27853, *Proteus vulgaris* ATCC 4636, *Bacillus subtilis* ATCC 6633, *Candida albicans* ATCC653/885. The bacterial concentration

was 10⁷ CFU/ml (determined by McFarland standard). Overnight cultures kept for 18–24 h at 36°C±1°C were used. The bacterial suspension was inoculated onto the entire surface of Mueller-Hinton agar (Dagestan Scientific Research Institute of Nutrient Media). The compounds were introduced to the wells in the form of DMSO solution in the concentrations of 100 µg/ml; the open wells were filled with 0.3 ml of the solution.

CONCLUSIONS

By the reaction of 3-phenyl-6-(α-bromoacetyl)-5-methylthieno[2,3-*d*]pyrimidine-2,4(1*H*,3*H*)-dione with thioacetamide 5-methyl-6-(2-methyl-1,3-thiazol-4-yl)-3-phenylthieno[2,3-*d*]pyrimidine-2,4(1*H*,3*H*)-dione was obtained; further the compound was modified by alkylation of its position 1 with benzyl chlorides and chloro-acetamides. 5-Methyl-6-(2-methyl-1,3-thiazol-4-yl)-3-phenylthieno[2,3-*d*]pyrimidine-2,4(1*H*,3*H*)-dione showed the antimicrobial activity against *Staphylococcus aureus* higher than the reference drugs Metronidazole and Streptomycin, it also appeared to be moderately active against *Pseudomonas aeruginosa* and *Candida albicans* fungi. For 1-alkyl-5-methyl-6-(2-methyl-1,3-thiazol-4-yl)-3-phenylthieno[2,3-*d*]pyrimidine-2,4(1*H*,3*H*)-diones the lower antimicrobial activity was found than that for the compound with unsubstituted position 1; the highest activity in this range was exhibited by compound **3b** with 4-methylbenzyl substituent at position 1. It moderately inhibits the growth of *Staphylococcus aureus* and *Candida albicans*.

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СИНТЕЗ ТА АНТИМІКРОБНА АКТИВНІСТЬ 5-МЕТИЛ-6-(2-МЕТИЛ-1,3-ТИАЗОЛ-4-ІЛ)-3-ФЕНИЛТІЕНО[2,3-*d*]ПІРИМИДИН-2,4(1Н,3Н)-ДІОНІВ

С.В.Власов, Т.П.Осолодченко, С.М.Коваленко, В.П.Черних

Ключові слова: тіофен; піримідин; алкілювання; тiazол

Шляхом реакції 3-феніл-6-(а-бромацетил)-5-метилтіено[2,3-*d*]піримідин-2,4(1Н,3Н)-діону з тіоацетамідом отримано 5-метил-6-(2-метил-1,3-тиазол-4-іл)-3-фенілтіено[2,3-*d*]піримідин-2,4(1Н,3Н)-діон, який далі модифікували шляхом алкілювання положення 1 бензилхлоридами та хлороацетамідами. Структура одержаних сполук була підтверджена даними ^1H ЯМР та мас-спектроскопії. Для одержаних сполук у спектрах ^1H ЯМР спостерігається сигнал протону тiazолу біля 7.62-7.67 м.ч., для 1-алкіл-5-метил-6-(2-метил-1,3-тиазол-4-іл)-3-фенілтіено[2,3-*d*]піримідин-2,4(1Н,3Н)-діонів присутні сигнали протонів метиленових груп у діапазоні 5.13-5.21 м.ч. для 1-бензилпохідних та при 4.78-4.82 м.ч. для сполук із ацетамідним фрагментом, для останніх також характерні чіткі сигнали протонів NH в діапазоні 9.68-10.41 м.ч. 5-Метил-6-(2-метил-1,3-тиазол-4-іл)-3-фенілтіено[2,3-*d*]піримідин-2,4(1Н,3Н)-діон виявив значну антимікробну активність, яка перевищила препарати порівняння метронідазол та стрептоміцин по відношенню до штаму *Staphylococcus aureus*, також він показав помірну пригнічуочу активність до росту *Pseudomonas aeruginosa* та грибів *Candida albicans*. Антимікробна активність 1-алкіл-5-метил-6-(2-метил-1,3-тиазол-4-іл)-3-фенілтіено[2,3-*d*]піримідин-2,4(1Н,3Н)-діонів поступається активності сполуки із незаміщеним положенням 1; найбільшу активність у цьому ряду проявила сполука, яка містить 4-метилбензильний замісник у положенні 1; вона помірно пригнічує рост *Staphylococcus aureus* та *Candida albicans*.

СИНТЕЗ И ПРОТИВОМИКРОБНАЯ АКТИВНОСТЬ 5-МЕТИЛ-6-(2-МЕТИЛ-1,3-ТИАЗОЛ-4-ИЛ)-3-ФЕНИЛТИЕНО[2,3-*d*]ПИРИМИДИН-2,4(1Н,3Н)-ДИОНОВ

С.В.Власов, Т.П.Осолодченко, С.Н.Коваленко, В.П.Черных

Ключевые слова: тиофен; пиримидин; алкилирование; тiazол

Путем реакции 3-фенил-6-(а-бромацетил)-5-метилтиено[2,3-*d*]пиримидин-2,4(1Н,3Н)-диона с тиоацетамидом получен 5-метил-6-(2-метил-1,3-тиазол-4-ил)-3-фенилтиено[2,3-*d*]пиримидин-2,4(1Н,3Н)-діон, который в дальнейшем модифицировали с помощью алкилирования положения 1 бензилхлоридами и хлорацетамидами. Структура полученных соединений была подтверждена данными ^1H ЯМР и масс-спектроскопии. Для полученных соединений в спектрах ^1H ЯМР наблюдается сигнал протона тiazола при 7.62-7.67 м.д., для 1-алкіл-5-метил-6-(2-метил-1,3-тиазол-4-іл)-3-фенілтиено[2,3-*d*]піримідин-2,4(1Н,3Н)-діонов присутствуют сигналы протонов метиленових групп в диапазоне 5.13-5.21 м.д. для 1-бензил производных и при 4.78-4.82 м.д. для соединений с ацетамідним фрагментом, для них также характерны четкие сигналы протонов NH в диапазоне 9.68-10.41 м.д. 5-Метил-6-(2-метил-1,3-тиазол-4-іл)-3-фенілтиено[2,3-*d*]піримідин-2,4(1Н,3Н)-діон проявил значительную противомикробную активность, которая превысила препараты сравнения метронидазол и стрептомицин по отношению к штамму *Staphylococcus aureus*, также он показал умеренную подавляющую активность роста *Pseudomonas aeruginosa* и грибов *Candida albicans*. Противомикробная активность 1-алкіл-5-метил-6-(2-метил-1,3-тиазол-4-іл)-3-фенілтиено[2,3-*d*]піримідин-2,4(1Н,3Н)-діонов уступает активности соединения с незамещенным положением 1; наибольшую активность в этом ряду проявило соединение, которое содержит 4-метилбензильный заместитель в положении 1; оно умеренно угнетает рост *Staphylococcus aureus* и *Candida albicans*.